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Sorbonne University M1 Chemistry- Physics et Chemistry of Materials (PCMat) Internship 2022- from April 15th to August 31st - Okayama University - JAPAN



## Micro/nano structures and color tones of hematite nano-particles for beautiful, highly bright, sustainable and heat-resistant red pigments

This internship takes place in the laboratory of

Professor FUJII Tatsuo, in the department of Applied Chemistry in the Graduate School of Natural Science and Technology-Okayama University.

The topic is focused on the improvement of red pigments, and more specifically on hematite,  $\alpha$ -Fe<sub>2</sub>O<sub>2</sub>, which is the most common red pigment used since ages, likely because hematite is the most stable iron oxide on earth.

In Japan, pigments are part of the paintings culture especially in Ukiyo-e paintings. Red pigment and more precisely hematite has been particularly present over ages. It can be found into BIZEN wares with red-brownish shades, but also in Fukiya

village, a remote rural district in the hills of Takahashi City, about 100kilometers north of Okayama, where the red color of walls and roofs is due to hematite. Pigments are called Bengara pigments.

The colour of hematite is red, sometimes deep dark and tends to tarnish over time. Here, one aim is to get highly bright and sustainable red pigment so that it can be applied in paintings. The paintings should also be heat-resistant not only to resist high temperatures during Japan's summer, but also in a vision of using less energy, especially regarding the Global Warming which is a current issue.

To achieve that goal, it was planned to:

- a modulate the hematite composition by doping it with Al or Si
- b adapt synthesis conditions and so the shape of hematite particles.

When pigments particles are synthetized, there are analyzed thanks to different techniques, such as X-ray diffraction (DRX), Scanning

electron microscopy (SEM), X-ray fluorescence (XRF), a spectrophotometer, Optical reflectance, and so on. Each technique is necessary to determine and check the composition, the shape of particles, the size of crystallites, but also to have the lattice parameters, and information about colors (redness, yellowness, brightness).



For that matter, Al is added to hematite to lighten up the color of the sample. Two different techniques have been used to synthetize the pigments particles. First a precipitation method was used and the particles shape obtained was fibrous. There, samples were either heated to 800°C either to 1000°C. In these two cases, we obtained quite different colors for same composition of Al. The second method used is the ultrasonic spray pyrolysis (USP), which consists in making spherical particles thanks to ultrasonic waves for different compositions, as did before. Here, an impact of the particles shape is also observed. Upon increasing the Al amount which lighten the



*Range of composition: From 0 Al to 0,5 Al added, obtained at 1000°C* 



Gathering with students of the laboratory

hematite colour, the size of the crystallite size decreases. The lattice constants also decreased, so the exchange integral between atomic orbitals increased. Then the bandgap is expected to get larger when the amount of Al gets bigger. The spherical shape was increasing the bandgap of hematite samples.



Si was also added to the composition. There, the brighter sample was obtained for a molar ratio of 0,1 Si. Samples were less pure with Si as well but the reflectance in visible light was very high. All brighter samples were tested into paintings, and it appeared that the highest reflectance into the infra-red region was obtained for 0,33 Al heated at 800°C. This was the most promising painting for application as heat-resistant pigment.



done with the brightest samples



*c) Painting done with 0,1 Si heated at* 800°*C*